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# Structure of Tris(phenylseleno)phosphine 

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#### Abstract

C}_{18} \mathrm{H}_{15} \mathrm{PSe}_{3}, \quad M_{r}=499.17, \quad\) trigonal (hexagonal cell), $\quad R \overline{3}, \quad a=12.8896$ (5), $\quad c=$ 19.1855 (7) $\AA, \quad V=2760.5$ (2) $\AA^{3}, \quad Z=6, \quad D_{x}=$ $1.80 \mathrm{~g} \mathrm{~cm}^{-3}, \quad \lambda(\mathrm{Mo} K \alpha)=0.71069 \AA \quad$ (graphite monochromator), $\mu=60.16 \mathrm{~cm}^{-1}, F(000)=1440, T$ $=296 \mathrm{~K}, R=0.068$ for 998 observed reflections with $I>3 \sigma(I)$. The structure determination provides the first crystallographic data of the strain-free $\mathrm{P}(\mathrm{SeCR})_{3}$ unit. Distances: P-Se 2.271 (2), Se-C 1.925 (6) A; angles: $\mathrm{Se}-\mathrm{P}-\mathrm{Se} 96.6$ (1), $\mathrm{P}-\mathrm{Se}-\mathrm{C} 97.6$ (2) ${ }^{\circ}$. The individual molecules have crystallographically imposed threefold symmetry and pack in layers, possibly giving the structure applications in intercalation chemistry.


Experimental. The air-sensitive product was synthesized at 295 K under an Ar atmosphere from the


Fig. 1. ORTEPII (Johnson, 1976) drawing ( $50 \%$ probability ellipsoids) of $\mathrm{P}\left(\mathrm{SeC}_{6} \mathrm{H}_{5}\right)_{3}$.


Fig. 2. Packing diagram ( $a$ axis towards viewer, $c$ axis horizontal).

Table 1. Atomic coordinates and equivalent isotropic temperature factors $\left(\AA^{2} \times 10^{4}\right)$ with e.s.d.'s in parentheses

| $U_{\mathrm{eq}}=(1 / 3) \sum_{i} \sum_{j} U_{i j} a_{i}{ }^{*} a_{j}^{*} \mathbf{a}_{i} \cdot \mathbf{a}_{j}$. |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | $x$ | $y$ | $z$ | $U_{\text {eq }}$ |
| $\mathrm{P}(1)$ | 0.0000 | 0.0000 | 0.3975 (1) | 387 (7) |
| $\mathrm{Se}(1)$ | 0.11911 (6) | -0.05200 (6) | 0.33751 (4) | 480 (8) |
| C(1) | 0.2620 (5) | 0.0373 (6) | 0.3924 (3) | 377 (52) |
| C(2) | 0.2777 (6) | -0.0149 (6) | 0.4524 (4) | 514 (63) |
| C(3) | 0.3826 (7) | 0.0472 (7) | 0.4905 (4) | 573 (72) |
| C(4) | 0.4717 (6) | 0.1578 (7) | 0.4687 (4) | 522 (66) |
| C(5) | 0.4560 (7) | 0.2102 (6) | 0.4096 (4) | 494 (61) |
| C(6) | 0.3492 (6) | 0.1479 (6) | 0.3710 (4) | 441 (56) |

Table 2. Selected bond lengths ( $\AA$ ) and angles $\left({ }^{\circ}\right)$ with e.s.d.'s in parentheses
$\mathrm{Se}(1)^{\prime}$ is related to $\mathrm{Se}(1)$ by the threefold axis along $0,0, z$.

| $\mathrm{P}(1)-\mathrm{Se}(1)$ | $2.271(2)$ | $\mathrm{C}(2)-\mathrm{C}(3)$ | $1.387(10)$ |
| :--- | :---: | :--- | ---: |
| $\mathrm{Se}(1)-\mathrm{C}(1)$ | $1.925(6)$ | $\mathrm{C}(3)-\mathrm{C}(4)$ | $1.374(11)$ |
| $\mathrm{C}(1)-\mathrm{C}(6)$ | $1.365(9)$ | $\mathrm{C}(4)-\mathrm{C}(5)$ | $1.386(11)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)$ | $1.398(10)$ | $\mathrm{C}(5)-\mathrm{C}(6)$ | $1.409(10)$ |
| $\mathrm{Se}(1)-\mathrm{P}(1)-\mathrm{Se}(1)^{\prime}$ | $96.6(1)$ | $\mathrm{C}(3)-\mathrm{C}(2)-\mathrm{C}(1)$ | $119.4(6)$ |
| $\mathrm{P}(1)-\mathrm{Se}(1)-\mathrm{C}(1)$ | $97.6(2)$ | $\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{C}(2)$ | $120.4(7)$ |
| $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{Se}(1)$ | $119.2(5)$ | $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | $120.3(7)$ |
| $\mathrm{C}(6)-\mathrm{C}(1)-\mathrm{Se}(1)$ | $120.1(5)$ | $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | $119.4(6)$ |
| $\mathrm{C}(6)-\mathrm{C}(1)-\mathrm{C}(2)$ | $120.6(6)$ | $\mathrm{C}(1)-\mathrm{C}(6)-\mathrm{C}(5)$ | $119.8(6)$ |

reaction of $2 \mathrm{M} \quad \mathrm{PCl}_{3} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ (Aldrich) with $\mathrm{NaSeC}_{6} \mathrm{H}_{5}$ (Liotta, Markiewicz \& Santiesteban, 1977) in THF. The mixture was filtered and clear yellow crystals (hexagonal plates) were obtained by slow solvent evaporation. A $0.61 \times 0.52 \times 0.30 \mathrm{~mm}$ capillary-mounted crystal was selected for data collection using a Huber (Crystal-Logic automated) four-circle diffractometer with $\theta / 2 \theta$ scan mode to a maximum $2 \theta$ of $50^{\circ}$ and $2 \theta$ scan speed of $3^{\circ} \mathrm{min}^{-1}$. Lattice parameters were determined from 41 reflections in the range $5.6<2 \theta<22.2^{\circ}$. Empirical absorption correction; maximum/minimum transmission factors were 1.00/0.288 (North, Phillips \& Mathews, 1968). Maximum $\sin \theta / \lambda=0.595 \AA^{-1} ; h, k$, $l$ range: $-7-15,-7-15,0-22$. Three standard reflections measured after every 97 reflections showed no significant variation in intensity $\left[\left(I_{\max }-I_{\text {min }}\right) / I_{\mathrm{av}}=\right.$ $0.021]$. 1243 reflections were measured, of which 1091 were unique ( $R_{\text {int }}=0.051$ ) and 93 were unobserved reflections with $I<3 \sigma(I)$. SHELXS86 (Sheldrick, 1985) direct-methods program was used to solve the
structure with $\mathbf{P}$ constrained on the threefold axis. Anisotropic (non-H atoms) full-matrix least-squares refinement; $\quad \sum w\left|\left|F_{o}\right|-\left|F_{c}\right|^{2}\right.$ minimized where $w=$ $1 /\left[\sigma\left(F_{o}\right)\right]^{2}$. The H -atom positions were calculated ( $\mathrm{C}-\mathrm{H}$ bond length $1.0 \AA$ ) and included as fixed contributors with isotropic thermal parameters fixed to $5.0 \AA^{2}$. 67 parameters were refined; $R=0.068, w R$ $=0.083, \quad S=3.23 ; \quad(\Delta / \sigma)_{\max }=0.017 ; \quad \Delta \rho_{\max / \min }=$ $0.84 /-1.00 \mathrm{e} \AA^{-3}$. Scatering factors were taken from International Tables for X-ray Crystallography (1974, Vol. IV). The UCLA Crystallographic Program Package (Strouse, 1985) was used throughout. The program PLOTMD (Luo, Ammon \& Gilliland, 1989) was used to modify the labels of the ORTEP drawing (Johnson, 1976) displayed in Fig. 1. Positional parameters and isotropic temperature factors are listed in Table 1; selected bond lengths and angles are listed in Table 2.* In Fig. 2, the crystal packing is displayed.

[^0]Related literature. The isomorphous structures $\mathrm{P}\left(\mathrm{SC}_{6} \mathrm{H}_{5}\right)_{3}$ (Burford, Royan \& White, 1990) and $\mathrm{As}\left(\mathrm{SC}_{6} \mathrm{H}_{5}\right)_{3}$ (Papalardo, Chakravorty, Irgolic \& Meyers, 1983) have been reported.

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# Structure of 3,3-Bis(2-imidazolyl)propionic Acid Monohydrate 

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#### Abstract

C}_{9} \mathrm{H}_{10} \mathrm{~N}_{4} \mathrm{O}_{2} \cdot \mathrm{H}_{2} \mathrm{O}, M_{r}=224.2\), triclinic, $P \overline{1}$, $a=7.322$ (1), $\quad b=10.029$ (1), $\quad c=7.155$ (1) $\AA, \quad \alpha=$ $89.96(1), \quad \beta=99.72(1), \quad \gamma=95.14(1)^{\circ}, \quad V=$ 515.8 (4) $\AA^{3}, Z=2, D_{x}=1.44 \mathrm{Mg} \mathrm{m}^{-3}, \lambda(\mathrm{Mo} K \alpha)=$ $0.71073 \AA, \mu=0.1 \mathrm{~mm}^{-1}, F(000)=236, T=295 \mathrm{~K}$, final $R=0.039$ for 1081 reflections. The molecule, abbreviated as HBIP, is a zwitterion containing $-\mathrm{COO}^{-}$and -(imidazole) $\mathrm{H}^{+}$residues. The

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dihedral angle between the two imidazole rings is $66.7(1)^{\circ}$. There is no intramolecular hydrogen bond.

Experimental. Synthesis according to Joseph, Leigh \& Swain (1977), colourless data-collection crystal of dimensions $0.35 \times 0.20 \times 0.08 \mathrm{~mm}$. $D_{m}$ not measured. Enraf-Nonius CAD-4 diffractometer; graphitemonochromated Mo $K \alpha$; cell dimensions from setting angles of 25 reflections having $10.6<\theta<13.1^{\circ}$; 1813 reflections measured using $\omega-2 \theta$ scan with $2 \theta$
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[^0]:    * Lists of structure factors, anisotropic thermal parameters, H-atom coordinates and bond lengths and angles have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 54971 (12 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CHI 2HU, England. [CIF reference: HH0566]

